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Key indicators

Single-crystal X-ray study T = 294 K Mean σ (C–C) = 0.004 Å R factor = 0.048 wR factor = 0.135 Data-to-parameter ratio = 14.9

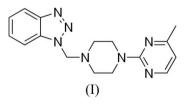
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

1-[4-(4-Methylpyrimidin-2-yl)piperazin-1-ylmethyl]-1*H*-benzotriazole

The title compound, $C_{16}H_{19}N_7$, is a potent new herbicide. X-ray analysis reveals that the piperazine ring adopts a chair conformation and weak $C-H \cdots N$ hydrogen bonds link the molecules into a chain along the *a* axis. Received 4 July 2005 Accepted 11 July 2005 Online 16 July 2005

Comment

Based on the reported 1.65 Å high-resolution crystal structure of spinach KARI (ketol-acid reductoisomerase) complex (Biou *et al.*, 1997), we obtained 279 molecules with low binding energy toward KARI from MDL/ACD three-dimensional database searching, using the program DOCK 4.0 (Wang *et al.*, 2004). These potential structures provide information for further design of new targeted KARI herbicidal molecules. According to the structural information and bioactivity data of benzotriazole, one of the 279 molecules provided by MDL/ACD three-dimensional database searching, a series of benzotriazole derivatives has been designed and synthesized. The X-ray crystal structure determination of the title compound, (I), was undertaken to investigate the relationship between structure and herbicidal activity.



The molecular structure of (I) is shown in Fig. 1. The X-ray analysis reveals that the piperazine ring is in a chair conformation. The C12—N5 distance of 1.370 (2) Å is shorter than the normal C—N single-bond distance of 1.47 Å (Carey, 2000), which shows that C12—N5 is conjugated with the pyrimidine ring. The molecules translated one unit cell along the *a*-axis direction are linked by weak C—H···N hydrogen-bonding interactions to form a chain (Fig. 2 and Table 2).

Experimental

Compound (I) was prepared according to the reported procedure of Bachman & Heisey (1946), using benzotriazole (0.01 mol), 4-methyl-2-(piperazin-1-yl)pyrimidine (0.11 mol) (Xu *et al.*, 1993), 40% formalin (0.012 mol) and methanol (15 ml) (2.47 g, 80% yield). Colourless single crystals suitable for X-ray diffraction analysis were obtained by recrystallization from ethanol.

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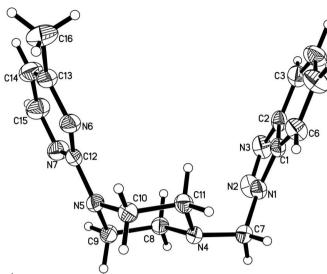


Figure 1

The molecular structure of (I), showing displacement ellipsoids drawn at the 30% probability level.

Z = 2

 $D_x = 1.334 \text{ Mg m}^{-3}$ Mo K α radiation Cell parameters from 1635 reflections $\theta = 2.3-26.2^{\circ}$ $\mu = 0.09 \text{ mm}^{-1}$ T = 294 (2) K Prism, colourless $0.26 \times 0.24 \times 0.20 \text{ mm}$

3104 independent reflections 2102 reflections with $I > 2\sigma(I)$

 $\begin{aligned} R_{\rm int} &= 0.018\\ \theta_{\rm max} &= 26.5^\circ \end{aligned}$

 $\begin{array}{l} h=-8\rightarrow8\\ k=-11\rightarrow11 \end{array}$

 $l = -16 \rightarrow 7$

Crystal data

C16H19N7
$M_r = 309.38$
Triclinic, P1
a = 6.5207 (17) Å
$b = 9.2488 (16) \text{\AA}$
c = 13.429 (4) Å
$\alpha = 102.108 \ (8)^{\circ}$
$\beta = 94.112 \ (7)^{\circ}$
$\gamma = 101.504 \ (8)^{\circ}$
V = 770.5 (3) Å ³

Data collection

Bruker SMART CCD area-detector
diffractometer
φ and ω scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)
$T_{\min} = 0.972, T_{\max} = 0.983$
4365 measured reflections

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0614P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.048$	+ 0.1903P]
$wR(F^2) = 0.135$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.02	$(\Delta/\sigma)_{\rm max} = 0.001$
3104 reflections	$\Delta \rho_{\rm max} = 0.34 \text{ e } \text{\AA}^{-3}$
209 parameters	$\Delta \rho_{\rm min} = -0.18 \text{ e} \text{ \AA}^{-3}$
H-atom parameters constrained	

Table 1

Selected	geometric	parameters	(A,	°).

N1-N2	1.352 (2)	N5-C12	1.370 (2)
N1-C1	1.360 (2)	N5-C9	1.451 (2)
N1-C7	1.473 (2)	N5-C10	1.452 (2)
N2-N3	1.298 (3)	N6-C12	1.337 (2)
N3-C2	1.365 (3)	N6-C13	1.352 (3)
N4-C7	1.430 (2)	N7-C15	1.330 (3)
N4-C8	1.457 (2)	N7-C12	1.336 (3)
N4-C11	1.457 (2)		
N4-C7-N1	115.55 (15)	N6-C13-C16	115.8 (2)
N7-C12-N6	126.34 (19)	C14-C13-C16	123.0 (2)
N7-C12-N5	117.49 (17)		

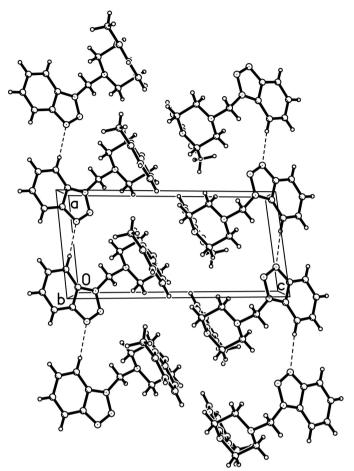


Figure 2

The crystal packing of (I), showing the C-H···N hydrogen-bonded chains (dashed lines).

Table 2

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
C6-H6···N3 ⁱ	0.93	2.50	3.430 (3)	176
$C9-H9A\cdots N7$	0.97	2.34	2.746 (3)	105
C10−H10B···N6	0.97	2.33	2.744 (3)	105

Symmetry code: (i) x + 1, y, z.

H atoms were placed in idealized positions and constrained to ride on their parent atoms, with C–H distances of 0.93 (aromatic), 0.97 (methylene) or 0.97 Å (methyl), and with $U_{\rm iso}(\rm H) = 1.2U_{eq}(\rm C)$ or $1.5U_{eq}(\rm methyl C)$.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1999); software used to prepare material for publication: *SHELXTL*.

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References

- Bachman, G. B. & Heisey, L. V. (1946). J. Am. Chem. Soc. 68, 2496-2499.
- Biou, V., Dumas, R. & Cohen-Addad, C. (1997). EMBO J. 16, 3405-3415.
- Bruker (1998). SMART. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (1999). SAINT and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
- Carey, F. A. (2000). Organic Chemistry, 4th ed. p. 861. New York: McGraw-Hill.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Wang, B.-L., Li, Z.-M. & Ma, Y., Wang, J.-G., Luo, X.-M. & Zuo, Z.-L. (2004). *Chin. J. Org. Chem.* 24, 973–976. (In Chinese.)
- Xu, Y., Zhu, Z.-H. & Tong, Z.-J. (1993). Chin. J. Pharm. 24, 49–51. (In Chinese.)